

# Why Quantitative XRD is a Valuable Tool

- XRF reports all Elements as Oxides
  - For example, all calcium in sample is reported as CaO
- XRD Reports Compounds per Crystalline Structures
  - For example, reports calcium as:
    - Calcite:  $\text{CaCO}_3$
    - Lime or Quick Lime:  $\text{CaO}$
    - Hydrated Lime or Portlandite:  $\text{Ca(OH)}_2$
    - Amorphous (no structure) Calcium:  $\text{Ca}$ 
      - Amorphous sulfur can be organic sulfur as S or fine precipitated sulfates that have not crystallized

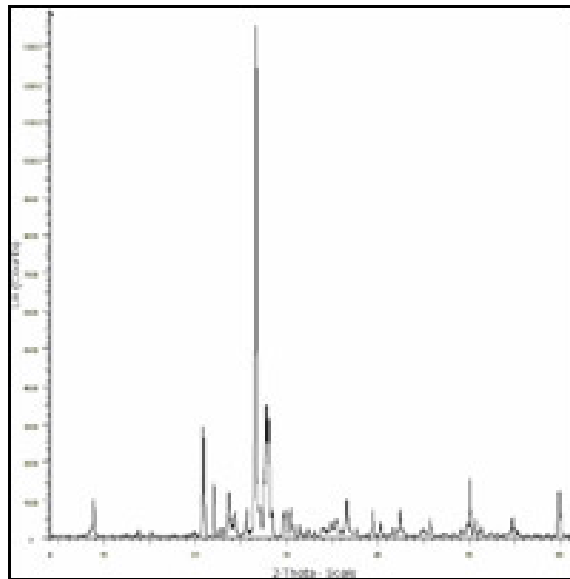
# Preparation Laboratory



PMET maintains a complete lab for analysis supporting the ash technology development and preparation of samples for SEM and XRD analysis.

# Quantitative XRD

Typical Results of XRD Analysis (Wt%)



| Exploration Drill Core |      |
|------------------------|------|
| quartz                 | 33.6 |
| k-feldspar             | 8.2  |
| plagioclase            | 23.4 |
| muscovite              | 14.4 |
| biotite                | 3.4  |
| chlorite               | 8.0  |
| kaolinite              | 5.0  |
| hematite               | 0.3  |
| pyrite                 | 1.6  |
| chalcopyrite           | 2.1  |

## XRD Analysis of Exploration Drill Core

Our XRD lab uses a Siemens D500 diffractometer with a Solex solid-state scintillation detector and state of the art data acquisition and processing software.

# Acid Mine Drainage Chemistry

- Not all forms of Sulfur Produce Acid Mine Drainage
- Formed via Oxidation of Pyrite:  
$$\text{FeS}_2 + \frac{15}{4} \text{O}_2 + \frac{7}{2} \text{H}_2\text{O} \rightarrow \text{Fe}(\text{OH})_3 + 2 \text{H}_2\text{SO}_4$$
- Quantification of  $\text{FeS}_2$  is Key Factor
- How to determine quantity of sulfur present as  $\text{FeS}_2$ ?

Answer:

## Quantitative XRD by Rietveld Refinement

- The Rietveld method is a whole pattern refinement based on instrument parameters, phase chemistry, and crystalline structure.
- A standard spike is used to calibrate system parameters and to calculate the concentration of non-crystalline material.

## Example Analysis of Major Brick Manufacturer Sample

- First Step, Measure Total Sulfur of the Samples

| <b>Total Sulfur Analysis</b> |          |
|------------------------------|----------|
| Sample A                     | Sample B |
| 0.120%                       | 0.122%   |

- Based on Sulfur Content Alone, Samples Appear Similar

## Example Analysis of Major Brick Manufacturer Sample

| Compound    | Sample A % | Sample B % |
|-------------|------------|------------|
| pyrite      | 0.10       | 0.00       |
| anhydrite   | 0.2        | 0.1        |
| jarosite    | 0.1        | 0.3        |
| hematite    | 0.1        | 0.7        |
| quartz      | 33.9       | 27.5       |
| K-feldspar  | 2.3        | 2.8        |
| plagioclase | 2.4        | 4.0        |
| muscovite   | 8.7        | 11.2       |
| clinochlore | 0.1        | 1.4        |
| calcite     | 1.0        | 9.3        |
| dolomite    | 0.2        | 0.6        |
| goethite    | 0.0        | 0.3        |
| kaolinite   | 28.3       | 7.2        |
| Amorphous   | 22.6       | 34.6       |

- Second step is analysis by XRD
- XRF analysis of this sample would report:
  - Pyrite as  $\text{SO}_3$  &  $\text{Fe}_2\text{O}_3$
  - Anhydrite as  $\text{CaO}$ ,  $\text{SO}_3$
  - Jarosite as  $\text{K}_2\text{O}$ ,  $\text{Fe}_2\text{O}_3$ , &  $\text{SO}_3$
  - Hematite as  $\text{Fe}_2\text{O}_3$
- XRD directly measures the amount of Pyrite

## Example Analysis of Major Brick Manufacturer Sample

|  |   | Sample A      | Sample B      |
|--|---|---------------|---------------|
| <b>Total Sulfur</b>                    |   | <b>0.120%</b> | <b>0.122%</b> |
| <b>Quantitative XRD</b>                |   |               |               |
| <b>Pyrite</b>                          | <b>FeS<sub>2</sub></b>  | <b>0.10%</b>  | <b>0.00%</b>  |
| <b>Anhydrite</b>                       | <b>CaSO<sub>4</sub></b>   | 0.2%          | 0.1%          |
| <b>Jarosite</b>                        | <b>KFe<sup>+3</sup>(SO<sub>4</sub>)<sub>2</sub>(OH)<sub>6</sub></b> | 0.1%          | 0.3%          |
| <b>Sulfur Partitioning</b>             |   |               |               |
| <b>Total Sulfur as FeS<sub>2</sub></b> |   | <b>44.2%</b>  | <b>0%</b>     |
| Total Sulfur as Anhydrite              |   | 45.1%         | 19.3%         |
| Total Sulfur as Jarosite               |   | 10.7%         | 31.5%         |
| Total Sulfur in Amorphous              |   | 0%            | 49.2%         |

The quantity of sulfur in the form of pyrite is determined